

1 Planck's radiation law & Einstein coefficients

26/3 13-15

1.1-1.5

$\vec{E}, \vec{B}$

Consider an electromagnetic field in vacuum
Maxwell's eqs $\Rightarrow$

$$\nabla^2 \vec{E}(\vec{r}, t) = \frac{1}{c^2} \frac{\partial^2 \vec{E}(\vec{r}, t)}{\partial t^2}$$

$$\nabla \cdot \vec{E} = 0$$

& the value of $\vec{B}$ follows from

$$\nabla \times \vec{E}(\vec{r}, t) = -\frac{\partial \vec{B}}{\partial t}$$

Assume box with conducting walls $0 \leq x \leq L$

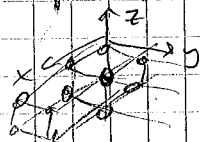
$\Rightarrow$ tangential components of $\vec{E}$ vanish

Sol: $\vec{E} = (E_x, E_y, E_z)$ where

$$\begin{cases} E_x(\vec{r}, t) = E_x(t) \cos(k_x x) \sin(k_y y) \sin(k_z z) \\ E_y(\vec{r}, t) = E_y(t) \sin(k_x x) \cos(k_y y) \sin(k_z z) \\ E_z(\vec{r}, t) = E_z(t) \sin(k_x x) \sin(k_y y) \cos(k_z z) \end{cases}$$

& $\vec{k} = (k_x, k_y, k_z)$ with $k_x = \frac{\pi}{L} m_x$, $m_x = 0, 1, 2, 3, \dots$
etc.

$\Rightarrow$ lattice of allowed wavevectors



(see Fig. 1.2)

Furthermore, M.E. $\Rightarrow \vec{k} \cdot \vec{E}(t) = 0$

where $\vec{E}(t) = (E_x(t), E_y(t), E_z(t))$

is the space-independent factor

For each $\vec{k}$, there are 2 linearly independent
directions for $\vec{E}(t)$, $\vec{E}_{E\lambda}(t)$ with $\lambda = 1, 2$

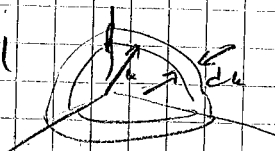
1-2

Each choice of $(\vec{k}, \lambda)$ is one field mode.

Frequency + direction of propagation + polarization

* How many field modes between $k, k+dk$

One octant of spherical shell



$$\frac{1}{8} \underbrace{4\pi k^2 dk}_{\text{shell volume}} \cdot \underbrace{\left(\frac{4}{\pi}\right)^{1/3}}_{\substack{\# \text{ } k\text{-values} \\ \text{per unit} \\ \text{volume}}} \cdot \underbrace{2}_{\text{polarization}}$$

↑
octant

Density of modes per unit volume

$$\rho(k) dk = \frac{k^2 dk}{\pi^2}$$

By using $\omega = ck$ we obtain

$$\rho(\omega) d\omega = \frac{\omega^2 d\omega}{\pi^2 c^3}$$

Planck's quantization hypothesis

Classically, the total energy

$$E = \frac{1}{2} \int dV (\epsilon_0 \vec{E}(r,t)^2 + \frac{1}{\mu_0} \vec{B}(r,t)^2)$$

$$(\epsilon_0 \mu_0 = c^2)$$

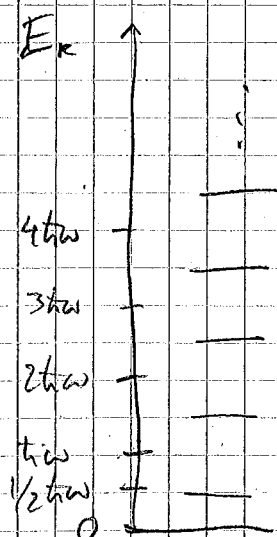
and $\vec{E}_0(t)$ can take on any continuous value.

→ instead, Planck suggested energy of each field mode can have any of the discrete values

$$E_R = (n + \frac{1}{2}) h\nu \quad n=0, 1, 2, 3, \dots$$

↑
zero-point energy!

n is the "number of photons" in each field mode
→ energy in a field mode can only change in discrete steps - quanta



allowed energies

$n=0$: $E_R = \frac{1}{2} h\nu$ the zero-point energy. Even though the field mode is empty.

$n > 0$: The field mode is "excited with n photons"
When n changes, photons are created or destroyed.

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Planck's radiation law

Assume a cavity in thermal equilibrium.

How many photons with frequency ω are excited at temperature T ?

Probability that a mode has energy E_n is given by the Boltzmann factor

$$P(n) = \frac{e^{-E_n/k_B T}}{\sum_n e^{-E_n/k_B T}} \quad E_n = (n + \frac{1}{2})\hbar\omega$$

$$= \frac{e^{-n\frac{\hbar\omega}{k_B T}}}{\sum_n e^{-n\frac{\hbar\omega}{k_B T}}} \quad (\frac{1}{2}\hbar\omega \text{ cancels out})$$

$$= \frac{U^n}{\sum_{n=0}^{\infty} U^n} \quad \text{where } U = e^{-\frac{\hbar\omega}{k_B T}}$$

$$\{\text{geom}\} = U^n(1-U)$$

Mean number of photons in this mode

$$\langle n \rangle = \sum_{n=0}^{\infty} n P(n) = (1-U) \sum_{n=0}^{\infty} n U^n = (1-U) U \frac{\partial}{\partial U} \sum_n U^n$$

$$= (1-U) U \frac{\partial}{\partial U} \frac{1}{1-U} = \frac{U}{1-U}$$

$$= \frac{1}{e^{\hbar\omega/k_B T} - 1}$$

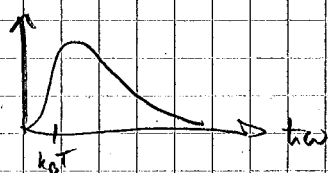
{ The Bose-Einstein dist. fun }

Calculate the energy density of the radiation in the cavity

$$\langle W_r(\omega) \rangle d\omega = \{ \# \text{ modes} \} \cdot \{ \# \text{ photons per mode} \} \cdot \{ \text{energy per photon} \}$$

$$= \frac{\omega^2}{\pi^2 c^3} d\omega \frac{1}{e^{h\omega/k_B T} - 1} h\omega$$

$$= \frac{h\omega^3}{\pi^2 c^3} \frac{1}{e^{h\omega/k_B T} - 1} d\omega$$



The black-body spectrum.

Planck 1900: the quantization hypothesis was only way to reproduce the empirically measured spectrum.

Fluctuations

Photons are continually created & destroyed.
Even if $\langle n \rangle = \text{constant}$ in equilibrium,
there are still fluctuations.

Ergodicity: A time average is the same
as the average taken over a large number
of identical systems (ensemble average)

Write

$$\langle n \rangle = \frac{U}{1-U}$$

$$\Rightarrow U = \frac{\langle n \rangle}{1 + \langle n \rangle}$$

then

$$P(n) = \frac{\langle n \rangle^n}{(1 + \langle n \rangle)^{1+n}}$$

Variance of photon number

$$(\Delta n)^2 = \langle n^2 \rangle - \langle n \rangle^2 = \sum_n (n - \langle n \rangle)^2 P(n)$$

Standard deviation

$$\Delta n = \sqrt{\langle n^2 \rangle - \langle n \rangle^2}$$

can be shown to be (look)

$$\Delta n = \sqrt{\langle n \rangle^2 + \langle n \rangle}$$

Note: 1) If $\langle n \rangle \gg 1 \Rightarrow \Delta n \approx \sqrt{\langle n \rangle}$

2) Always $\Delta n > \langle n \rangle$, Wide-spread distribution.

Fig 1.7



Einstein's theory of absorption & emission

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N atoms interacting with an EM field.

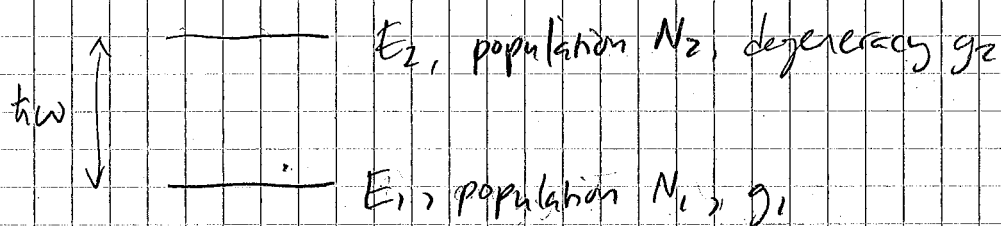
A lot can be done without applying a fully QM theory, just 2 assumptions:

- 1) Atoms have quantized energy levels
- 2) Light is created & destroyed in quanta.

Assume N identical atoms

with two energy levels

(only these two matter - explain later)



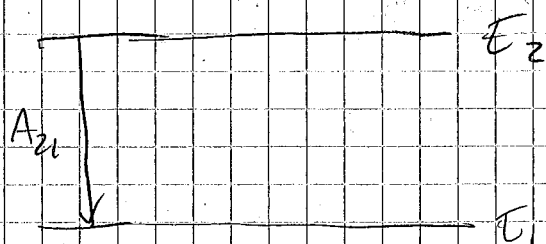
assume $E_2 - E_1 = h\omega$

and $N_1 + N_2 = N$

The density of radiation energy

$$\langle W(\omega) \rangle = \underbrace{\langle W_T(\omega) \rangle}_{\text{thermal}} + \underbrace{\langle W_E(\omega) \rangle}_{\substack{\text{external source} \\ \text{(beam of light)}}}$$

Energy exchange:



Spontaneous emission rate A_{21} $[s^{-1}]$
prob. per unit time

Absorption $\langle W(\omega) \rangle B_{12}$ $[s^{-1}]$

Stimulated emission $\langle W(\omega) \rangle B_{21}$ $[s^{-1}]$

The "Einstein coefficients",

Rate equations for the level populations

$$\left[\begin{aligned} \frac{dN_1}{dt} &= N_2 A_{21} - N_1 B_{12} \langle W(\omega) \rangle + N_2 B_{21} \langle W(\omega) \rangle \\ \frac{dN_2}{dt} &= - \frac{dN_1}{dt} \end{aligned} \right]$$

Steady state solution:

$$\text{Put } \frac{dN_1}{dt} = \frac{dN_2}{dt} = 0$$

$$N_2 A_{21} - N_1 B_{12} \langle W(\omega) \rangle + N_2 B_{21} \langle W(\omega) \rangle = 0$$

Now assume thermal equilibrium: $\langle W_E(\omega) \rangle = 0$

We can calculate the Einstein coeffs.

$$\langle W_T(\omega) \rangle (B_{21} N_2 - B_{12} N_1) = -A_{21} N_2$$

$$\langle W_T(\omega) \rangle = \frac{A_{21}}{\frac{N_1}{N_2} B_{12} - B_{21}}$$

Use thermal equ. $\Rightarrow$ Boltzmann distribution

$$\frac{N_1}{N_2} = \frac{e^{-E_1/k_B T}}{e^{-E_2/k_B T}} \cdot \frac{g_1}{g_2} = \frac{g_1}{g_2} e^{\hbar\omega/k_B T}$$

also use Planck's law

$$\langle W_T(\omega) \rangle = \frac{\hbar\omega^3}{\pi c^3} \frac{1}{e^{\hbar\omega/k_B T} - 1}$$

$$\frac{\hbar\omega^3}{\pi c^3} \frac{1}{e^{\hbar\omega/k_B T} - 1} = \frac{A_{21}}{\frac{g_1}{g_2} e^{\frac{\hbar\omega}{k_B T}} B_{12} - B_{21}}$$

this holds for all temperatures T if

$$\left\{ \begin{array}{l} \frac{g_1}{g_2} B_{12} = B_{21} \quad \Rightarrow \quad g_1 B_{12} = g_2 B_{21} \\ \frac{A_{21}}{B_{21}} = \frac{\hbar\omega^3}{\pi c^3} \end{array} \right.$$